

Effect of Dioxygen Exposure on Hydrogen Peroxide Production by *Bacillus subtilis* Oxalate Decarboxylase in the Absence of Substrate

Megan Booth and Dr. Alexander Angerhofer

Department of Chemistry, University of Florida, Gainesville FL 32611-7200, USA



Introduction

Oxalate decarboxylase (OxDC) from *Bacillus subtilis* is a manganese dependent enzyme that catalyzes the breakdown of oxalate into CO₂ and formate using bound O₂ as a cofactor.^{1,2} It is known that Mn(III) drives the reaction; however, the curious dual dependence on O₂ has garnered attention, including questions such as

- Where does O₂ bind?
- Is there a LRET process involved between bound O₂ and the catalytically active Mn?
- Is O₂ only required to generate Mn(III) or does it actively participate in decarboxylation?

The answer to this first question awaits further study; however, new evidence presented herein supports the theory that O₂ binding drives Mn oxidation, acting as an electron sink to form superoxide and its acid catalyzed dismutation product H₂O₂



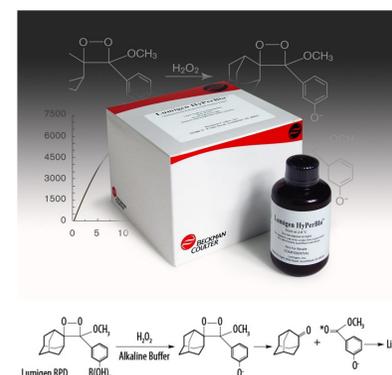
Unpublished crystal structure of wild type OxDC. Manganese centers are shown in purple. Monomer is denoted in blue.

Methodology

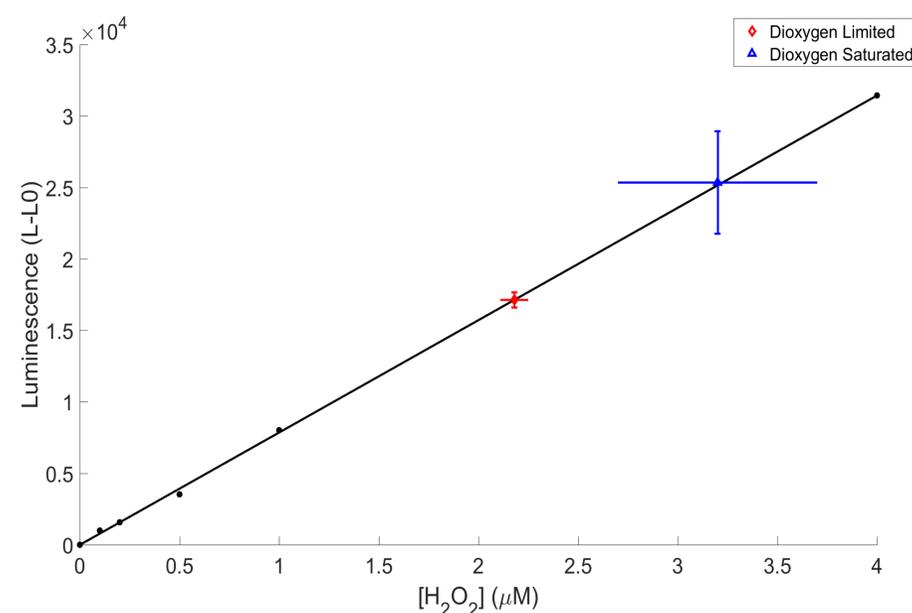
Enzyme expression, purification, and concentration

Anaerobic dialysis from pH 8.5 to 4.0 and reintroduction of O₂

HyPerBlu chemiluminescent assay to determine extent of H₂O₂ production



Results



	Dioxygen Limited	Dioxygen Saturated
Final [OxDC]	390 ± 30 μM	340 ± 30 μM
[H ₂ O ₂] Produced	2.18 ± 0.07 μM	3.22 ± 0.45 μM
[H ₂ O ₂]/[OxDC]	5.6 ± 0.5 (x10 ⁻³)	9.5 ± 1.5 (x10 ⁻³)
% Difference in [H ₂ O ₂]/[OxDC]	70%	

Conclusions

- A 70% increase in H₂O₂ production was observed upon saturating OxDC with O₂ suggesting a direct correlation between O₂ binding and Mn oxidation
- Future experiments will be focused on establishing a true baseline for H₂O₂ by ensuring O₂ is eliminated from the system. This will allow further studies to see if activity is solely dependent on Mn oxidation or if O₂ plays an active role in addition to facilitating oxidation

References

1. Tanner, A. et al. (2001) *J. Biol. Chem.* **276**: 43627-43634.
2. Twahir, U et al. (2016) *Biochemistry.* **55**: 6506-6516.

Funding

Funding for this project was provided by the University of Florida's University Scholars Program